

**Code No: 153BZ****JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD****B.Tech II Year I Semester Examinations, September - 2021****THERMODYNAMICS****(Mechanical Engineering)****Time: 3 hours****Max. Marks: 75****Answer any five questions****All questions carry equal marks**

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- 1.a) During one cycle the working fluid in an engine engages in two work interactions: 15 kJ to the fluid and 44 kJ from the fluid, and three heat interactions, two of which are known: 75 kJ to the fluid and 40 kJ from the fluid. Evaluate the magnitude and direction of the third heat transfer.
- b) Explain the significance of absolute scale of temperature. [8+7]
- 2.a) Using an engine of 30% thermal efficiency to drive a refrigerator having a COP of 5, what is the heat input into the engine for each MJ removed from the cold body by the refrigerator? If this system is used as a heat pump, how many MJ of heat would be available for heating for each MJ of heat input to the engine?
- b) Describe the corollaries of first law of thermodynamics. [8+7]
3. Water at 150 kPa and 10 °C enters a mixing chamber at a rate of 136 kg/min where it is mixed steadily with steam entering at 140 kPa and 115 °C. The mixture leaves the chamber at 140 kPa and 55 °C and heat loss to the surrounding air at 22 °C at a rate of 190 kJ/min. Neglecting KE and PE, determine the entropy generation, available energy and irreversibility. [15]
- 4.a) 0.45 kg of air undergoes two processes that are in series. The first process consists of an isothermal expansion during which the air is supplied with 23.2 KJ at a temperature of 187°C this process is then followed by a reversible adiabatic process that takes the gas to a temperature of 50°C, determine the change in entropy for these combined processes.
- b) Differentiate between unavailable energy and irreversibility. [8+7]
- 5.a) Steam having a quality of 0.80 is contained in a rigid vessel at a pressure of 20.7 bar. Heat is added until the temperature reaches to 260°C. Determine the final pressure.
- b) Explain in detail about a typical P – V – T diagram for a substance that expands on freezing. [8+7]
- 6.a) Two Carnot engines are operated in a series with the exhaust of the first engine being the input of the second engine. The upper temperature of this combination is 700 °C, and the lower temperature is 200°C if each engine has the same thermal efficiency, determine the exhaust temperature of the first engine and the inlet temperature to second engine.
- b) How to measure the dry bulb temperature and wet bulb temperature? Explain. [8+7]
- 7.a) Derive the Vander Waals equation for real gases.
- b) Explain the steps involved in the construction of Psychrometric chart at 2 bar pressure and also explain the process of adiabatic saturation. [7+8]
- 8.a) Derive an expression for the thermal efficiency of a dual cycle.
- b) Determine the mean effective pressure of a four cycle, six cylinder engine that has a 75 kW output at 4000 rpm. The cylinder bore is 75 mm and the stroke is 102 mm. [7+8]